

CONFIGURATION OF RETICULINE IN THE OPIUM POPPY

A.R. Battersby and G.W. Evans

The Robert Robinson Laboratories,
University of Liverpool, England.

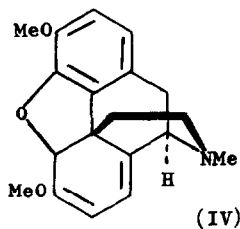
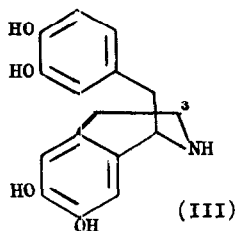
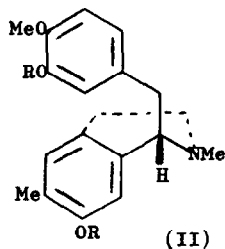
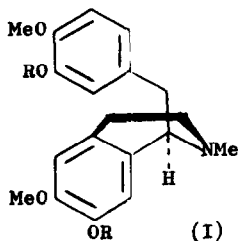
and

R.O. Martin, M.E. Warren, Jr. and H. Rapoport

Department of Chemistry, University of California,
Berkeley, California, U.S.A.

(Received 1 March 1965)

Labelled (\pm)-reticuline (as I, R=H) has been shown (1,2) to give rise in poppy plants to specifically labelled morphine alkaloids, e.g. thebaine (IV) and, further, (\pm)-reticuline has been isolated from crude opium (3). These results, when taken with the proof that (-)-reticuline (I, R=H) is the precursor of the morphine alkaloids in the living system (4,5) make it important to determine which epimer of reticuline predominates in opium poppy plants. To this end, (\pm)-[3- 14 C]norlaudanosoline (III) was fed over short feeding periods to Papaver somniferum plants (4); it is known that labelled norlaudanosoline gives rise to specifically labelled morphine alkaloids (2). The total alkaloids of the plants were isolated by a method which was proved to extract reticuline, and the further examination of this mixture is outlined below.



(-)-Reticuline (I, R=H) (4) was ethylated with diazoethane to give (-)-O₂O-diethylreticuline (I, R=Et), $[\alpha]_D^{20} -59.3^\circ$ (c, 1.0 in CHCl_3). The (+)-isomer (II, R=H) similarly afforded (+)-O₂O-diethylreticuline (II, R=Et), $[\alpha]_D^{20} +58.7^\circ$ (c, 1.0 in CHCl_3). A small quantity of (-)-O₂O-diethylreticuline (I, R=Et) multiply labelled with ^{14}C and ^3H was prepared by ethylation of the corresponding labelled reticuline (4) and this was mixed with a large excess of radio inactive (-)-O₂O-diethylreticuline (II, R=Et). The crystalline picrolonate prepared from this mixture was recrystallised several times and a steady fall in the radioactivity of the picrolonate occurred. After three crystallisations, ca. 5% of the original activity was retained. This establishes that dilution analysis is applicable to (+)- and (-)-O₂O-diethylreticulines in the following way.

The total alkaloids isolated above from the plants were divided into two halves. To one was added a known quantity of radio-inactive (+)-reticuline and the total mixture was O-ethylated with diazoethane. Chromatography of the products give radioactive O,O-diethylreticuline (9 mg.) which was diluted with radio-inactive (+)-O,O-diethylreticuline (60 mg.) and then crystallised several times as the picrolonate. A constant activity of 2.9×10^5 dis./100 sec./mmole. was reached after three crystallisation steps. The other half of the total alkaloids was treated in the same way, save that radio-inactive (-)-reticuline and (-)-O,O-diethylreticuline were substituted for the (+)-epimers at the appropriate points. Five crystallisations of the picrolonate were necessary to achieve a constant activity of 4.3×10^4 dis./100 sec./mmole. The pool of (+)-reticuline thus carries over six times as much activity as the pool of (-)-reticuline. This result is in agreement with the isolation from opium by Brochmann-Hanssen and Nielson of reticuline which has an excess of the (+)-form; their results are reported in the preceding paper (6).

It has recently been shown (5) that five-day old poppy seedlings grown for 2.5 hr. in ^{14}C -carbon dioxide produce highly active reticuline. Some of this material was purified by thin layer chromatography and examined by dilution analysis as above. The total activities of the (+)-, and (-)-O,O-diethylreticulines so obtained were found to be equal within experimental error. This is in clear contrast to the results from opium and from mature poppies fed norlaudanosoline. A possible explanation for this difference might be the rapid conversion of (-)-reticuline into morphine

alkaloids in flowering plants, as compared to a slower metabolism of the (+)-isomer. In seedlings, both isomers appear to be metabolized at an equal and rapid rate.

Acknowledgements. We (A.R.B. and G.W.E.) thank the Edinburgh Pharmaceutical Industries and the Department of Scientific and Industrial Research for financial support and also Mr. J. K. Hulme (Ness Botanic Gardens) for valuable help with the plants; supported (R.O.M., M.E.W., and H.R.) in part by the United States Atomic Energy Commission and Grant B-570 from the National Institutes of Health, United States Public Health Service.

References

1. A. R. Battersby, R. Binks, R. J. Francis, and H. Ramuz, J. Chem. Soc. 1964, 3600.
2. D. H. R. Barton, G. W. Kirby, W. Steglich, G. M. Thomas, A. R. Battersby, T. A. Dobson and H. Ramuz, J. Chem. Soc., 1965, 0000.
3. E. Brochmann-Hanssen and T. Furuya, J. Pharm. Sci., 53, 575, 1964.
4. A. R. Battersby, D. M. Foulkes and (in part) R. Binks, J. Chem. Soc., 1965, 0000.
5. R. O. Martin, M. E. Warren, Jr., and H. Rapoport, J. Amer. Chem. Soc., 1964, 86, 4726.
6. E. Bröchmann-Hanssen and B. Nielson, Tetrahedron Letters, 18, 1271, 1965